

CO-FIRING OF BIOMASS WITH GAS FUEL IN LOW-POWER BOILERS

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Abstract. Combustion is a complex process. It consists of many physical and chemical processes of thermal decomposition and fuel combustion taking place in a specific space and time. This paper examines the evaluation of the process of combustion pellets and briquettes made from oat straw in a low-power grate boiler 10 kW, which was carried out in two variants: without modification of the combustion chamber and with its modification. In the boiler processing, an element limiting the combustion chamber was used; it was made of steel strips with dimensions (length × width × thickness) of 100×3×1 mm attached radially at one end to a steel core 50 mm long, under which, using a mixture of butane and propane in a 70:30 ratio, a secondary combustion zone was created. Combustion of pellets without modification of the combustion chamber was characterized by increased dynamics, which was indicated by higher combustion rate, higher flue gas temperature and lower air excess. At the same time, modification of the combustion chamber applied in the scope of these parameters contributed to reduction of the combustion rate and flue gas temperature, while reducing the air excess that was absorbed by combustion of the gaseous fuel. Emission of pollutants with combustion of pellets and briquettes without modification of the combustion chamber was characterized by similar values in the CO and NO range, only for briquettes, the SO₂ emission was more than twice as high. However, by modifying the combustion chamber along with the use of additional gaseous fuel, a reduction of CO by 15 % was achieved. Higher NO and SO₂ emissions were observed. It would be expedient to conduct research to a greater extent related to optimization of CO, NO and SO₂ emissions using other types of herbaceous biomass.

Keywords: combustion, emission, herbaceous biomass.

Introduction

In recent years, Polish rural areas have undergone major structural, spatial and economic changes. Currently, shaping and development of these areas cannot exist without taking into account the protection of natural resources. Therefore, there are large needs in the field of environmental engineering and sustainable development of these areas [1; 2]. In this respect, in many countries, the research is conducted that focuses on: the quality of agricultural raw materials [3-8], new technologies in the field of agriculture using modeling techniques [9-11], safety of technical facilities, animals and the population [12-13], as well as in the field of energy, paying attention to renewable energy sources [14-16].

Research on the problem of the quality of the combustion process in low power heating devices powered by solid fuels is undertaken because of significant impact of such devices on air quality. In countries such as Poland, the share of individual heating is relatively significant. The popularity of such a solution results from the poorly developed gas fuel distribution system infrastructure and much lower solid fuel prices. At the same time, many thermal installations, especially the old ones, are supplied with coal [17-19].

Combustion is a complex process. It consists of many physical and chemical processes of thermal decomposition and fuel combustion, taking place in a specific space and time. The quality of the combustion processes in low-power boilers to about 20-25 kW results directly from the thermodynamic parameters of the combustion process in the boiler and construction of the combustion chamber.

Therefore, the analysis of the combustion process requires knowledge of the properties of fuels and their impact on its course. Less carbonization of biomass fuels, apart from affecting the technical characteristics of fuels, e.g. on calorific value, also determines the course of the combustion process in its individual phases [20-22].

Considering the phases of this process, we distinguish the following ones: drying, degassing, coke gasification and slag cooling processes, successively on the grate. It is important that after charging boilers into the combustion chamber a transient state arises, characterized by increased emission dynamics, which gradually goes into steady state until the flame disappears and the coking residue is

burnt. The disadvantage of grate boilers, especially those powered by biomass, is the free flow of volatile substances through the heat exchanger towards the chimney. This results in increased hydrocarbon emissions and thus reduction in the efficiency of the combustion process [19; 23].

Therefore, this research was carried out to assess the process of combustion pellets and briquettes from oat straw in a low-power grate boiler 10 kW, which was carried out in two variants: without modification of the combustion chamber and with its modification.

Materials and methods

Oat straw was selected for the research as a waste material that is widely available in connection with the production of oats. This biomass came from a farm located in Lublin province. Representative samples weighing 25 kg were taken from the field, then ground with a H-111 hammer mill with 10 mm sieves. Then it was agglomerated to pellets and briquettes. Pelleting was carried out in a granulator using an 8 mm matrix and briquetting in a hydraulic briquetting machine with a sleeve diameter of 50 mm.

When determining the physical and chemical properties of raw materials and produced biofuels, the following methods were used (measurements were carried out in triplicate):

- geometrical features (diameter and length) – according to EN ISO 17829: 2016-02 [24];
- moisture – weight method according to EN ISO 18134-3: 2016 [25];
- density – calculated on the basis of geometrical features in the drawn samples and the mass of biofuel portions, using the following formula:

$$\rho_w = \frac{4 \cdot 10^6 \cdot m}{\pi \cdot d^2 \cdot l} \quad (1)$$

where ρ_w – density of briquettes or pellets, $\text{kg} \cdot \text{m}^{-3}$,
 m – mass of the fuel portion, briquettes or pellets, g,
 d – average diameter of briquettes or pellets, mm,
 l – sum of the length of briquettes or pellets in a fuel portion, mm.

- carbon, hydrogen, nitrogen – determination according to EN ISO 16948: 2015 [26];
- sulfur – determination according to EN ISO 16994: 2015 [27];
- part of volatile components – determination according to EN ISO 18123: 2016 [28];
- net calorific value – calculation after determining the calorific value according to EN ISO 18125: 2017 [29];
- ash – according to the EN ISO 18122: 2016 standard [30].

The fuel collected for testing in portions of 1 kg was burned in a grate heating device, fueled periodically. Two combustion variants were adopted. The first one without modification of the combustion chamber, while in the second one at the top of the combustion chamber, a chamber limiting element made of steel sheet strips with dimensions (length \times width \times thickness) 100 \times 3 \times 1 mm attached radially to one end with a 50 mm long steel core was applied, under which a secondary combustion zone was created with a mixture of butane and propane in a ratio of 70:30 (Figure 1). The purpose of this zone was to additionally burn off volatile organic compounds, such as CO. Gas supply and flame were active for the initial 300 s of combustion the fuel portion.

Process timing, determinations of flue gas composition and temperature measurements were carried out continuously, i.e. from the moment of ignition to the time the reaction expired, for which the flue gas temperature drop to 120 °C was assumed. Flue gas was taken from the chimney. The measuring probe was connected to an exhaust gas dryer, from which the exhaust gas went to the exhaust gas analyzer. During the tests, a portable exhaust gas analyzer based on infrared sensors (NDIR) was used for the following gases: CO, CO₂, NO, SO₂ and electrochemical – O₂. The temperature was measured using a K-type thermocouple, which was located in the exhaust gas sampling probe. The results were automatically recorded in the analyzer database every 2 s, with the simultaneous recording of the data acquisition time.

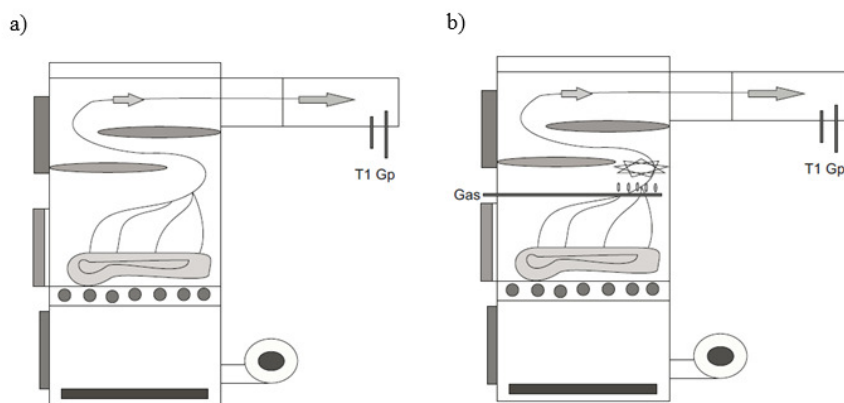


Fig. 1. **Diagram of the test stand (combustion chamber):** a) – before modification; b) – after modification; Gas – gas fuel fed to the combustion chamber, T1 – temperature measurement in the chimney, Gp – flue gas collection

Results and discussion

The average values of the results obtained (from three replicates) characterizing the raw material used and the biofuels made are presented in Tables 1 and 2.

Table 1

Technical and chemical properties of biomass of oat straw

Parameter	Symbol	Unit	Oat straw
Total moisture	W_i^r	%	10.833±0.016
Volatile meter	V^d	%	73.330±0.102
Net calorific value	Q_i^r	MJ·kg ⁻¹	15.475±0.031
Ash	A^d	%	6.455±0.125
Elemental composition	C^d	%	43.670±0.286
	H^d	%	5.223±0.039
	N^d	%	0.440±0.052
	S^d	%	0.007±0.002

Table 2

Physical properties of biofuels made of oat straw

Parameter	Symbol	Unit	Pellets	Briquettes
Length	L	mm	34±0.5	29±0.5
Diameter	D	mm	8±0.5	52±0.5
Density	–	kg·m ⁻³	1016±12	858±11
Mechanical durability	DU	%	98.17±0.25	97.57±0.21

The oat straw used during the tests was characterized by comparable technical and chemical characteristics with the results of the tests presented by Górnicki et al. [31]. Particularly noteworthy here is the content of volatiles at the level of 73 %, a relatively high calorific value comparable to other types of straw 15.47 MJ·kg⁻¹ and quite high ash content at the level of 6.5 % (Table 1). The improvement of geometrical parameters of this biomass is obtained in the process of its compaction. The pellets and briquettes produced during the tests in the range of considered features were characterized by desirable geometrical features, thus meeting the assumptions of relevant quality standards [20; 32].

The recorded parameters of the process of combustion pellets and briquettes are presented in Table 3. Combustion tests of fuels selected for the study were conducted under comparable conditions when fed with air stream from below under the grate, with an average rate of 2.5 m·s⁻¹. During the tests, the combustion rate for pellets was about 30 % higher than for briquettes and amounted to 4.36 kg·h⁻¹, while for briquettes this parameter was about 3 kg·h⁻¹.

Table 3

Parameters of the combustion process

Combustion system		Combustion rate, kg·h ⁻¹	Air excess coefficient	Exhaust gas temperature, °C
Pellets combustion made of oat straw	without gas fuel	4.36±0.10	5.73±3.13	260±112
	with gas fuel	4.20±0.08	5.53±3.22	207±70
Briquettes combustion made of oat straw	without gas fuel	3.10±0.07	5.98±2.19	174±32
	with gas fuel	2.68±0.07	5.68±1.99	163±38

Differences between fuel grades are also marked for the value of the air excess coefficient. Due to the constant air supply from the fan, during combustion of pellets, the air excess averaged to 5.5 (minimum about 2.5), while for briquettes – 6 (minimum about 3.5), but for pellets, larger fluctuations of this parameter were recorded, indicating the dynamics of combustion of this fuel. The modification of the boiler contributed to the reduction of the combustion rate and the air excess slightly decreased, obtaining in the combustion stabilized period the values close to those recommended in literature [19; 23]. Differences related to the type of assortment and the combustion system were also reflected in the flue gas temperature indications, which for pellets was 260 °C and 207 °C on average without and with boiler modification, whereas for briquettes, it was 163 °C and 174 °C (Table 3). The applied modification of the combustion chamber (when a steel element was used together with gas fuel) in the scope of these parameters contributed to the reduction of the combustion rate and temperature of the flue gas in the chimney, while reducing the air excess that was absorbed by the combustion of the gaseous fuel. The content in exhaust gases as well as CO, NO, SO₂, and CO₂ emission factors are presented in Table 4.

Table 4

Emissions CO, NO, SO₂, CO₂ of the exhaust during combustion of oat straw without and with gas fuel

Combustion system	Parameters	CO, ppm	CO, mg·m ⁻³ at 10 % O ₂	NO, ppm	NO, mg·m ⁻³ at 10 % O ₂	SO ₂ , ppm	SO ₂ , mg·m ⁻³ at 10 % O ₂	CO ₂ , %
Pellets without gas fuel	maximum	11240	88929	272	1209	113	2045	7.86
	average	4191	18829	118	421	20	235	3.75
	kurtosis	4.10	7.13	-0.45	14.61	13.23	11.06	-0.75
	skewness coefficient	0.70	2.27	0.21	3.36	3.28	3.24	0.16
Pellets with gas fuel	maximum	11160	57382	358	1787	290	2968	7.60
	average	4165	16033	201	616	72	623	4.61
	kurtosis	1.28	0.27	0.42	5.54	6.31	2.69	-0.72
	skewness coefficient	1.32	1.31	-0.64	2.48	2.66	1.93	-0.41
Briquettes without gas fuel	maximum	6770	43923	145	1147	124	1960	5.41
	average	4592	17431	118	463	76	678	3.74
	kurtosis	1.98	1.64	2.35	3.70	-0.12	1.30	-0.01
	skewness coefficient	-0.71	1.37	-1.36	1.70	0.45	1.32	-0.45
Briquettes with gas fuel	maximum	3733	37385	105	1260	75	1972	3.4
	average	2257	16588	66	489	48	799	1.94
	kurtosis	0.34	1.44	-0.20	9.21	0.23	3.32	-0.33
	skewness coefficient	-0.49	0.88	-0.51	1.69	-0.28	1.30	0.05

Under the test conditions, pellet combustion occurred most rapidly with significant CO, NO and SO₂ emissions. At the same time, the use of an additional flame zone resulted in 15 % reduction in the average value of emitted CO calculated as normal conditions at 10 % O₂, but it resulted in an increase of over 100 % in NO and SO₂ emissions. However, NO emission during pellet combustion in the second variant probably increases according to the Zeldovich mechanism. During the combustion of briquettes in the first adopted variant of combustion CO, NO emissions were comparable to the

emission of these compounds during pellet combustion, while the SO₂ emission was much higher. However, in the second variant with the modification of the combustion chamber, the same reduction by 15 % in the CO emission factor was achieved, also observing higher SO₂ emission with briquette combustion (Table 4). The values obtained in the first variant of combustion tests were comparable with the data presented by other researchers analyzing the emission of solid biofuels in the form of pellets and briquettes [18-20; 33]. However, it is puzzling, what needs verification during testing with other fuels, why more SO₂ emissions arise.

Conclusions

1. Combustion of pellets under the test conditions during the first variant without modification of the combustion chamber was characterized by increased dynamics, which was indicated by: higher combustion rate, higher flue gas temperature and lower air excess. At the same time, modification of the combustion chamber applied in the scope of these parameters contributed to reduction of the combustion rate and flue gas temperature, while reducing the air excess that was absorbed by the combustion of gaseous fuel.
2. Emission of pollutants with combustion pellets and briquettes without modification of the combustion chamber was characterized by similar emissions in the CO and NO range and more than twice as much SO₂ emissions for briquettes. However, by modifying the combustion chamber along with the use of additional gaseous fuel, CO reduction of 15 % was achieved. Higher NO and SO₂ emissions were observed.
3. It would be expedient to conduct research to a greater extent related to optimization of CO, NO and SO₂ emissions using other types of herbaceous biomass.

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